

ADDITION OF AMINE NUCLEOPHILES TO DIPHOSPHORYLALKYNES.

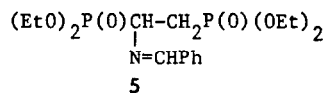
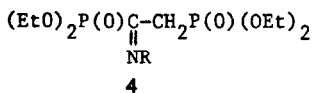
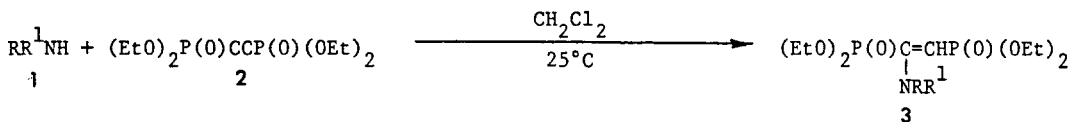
THE CHEMISTRY OF THE DERIVED ENAMINES

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Abstract: Amine nucleophiles bearing at least one hydrogen substituent add across tetraethyl ethynyldiphosphonate to give enamine derivatives, one of which (derived from cyclohexylamine) has been evaluated as an imine anion precursor in alkylations and Wittig-Horner-Emmons types of reactions.

We would like to report the synthesis and characterization of several members of a novel class of enamines 3, and to elucidate the synthetic potential of some of such species. Reaction of various amine nucleophiles 1 with tetraethyl ethynyldiphosphonate 2<sup>1,2</sup>, at room temperature in dichloromethane gave 3 in high yields (Table I).



Depending upon the nature of R and R<sup>1</sup>, products 4 or 5 might be obtained ultimately. Thus with R<sup>1</sup>=H and R=NH<sub>2</sub> or OH, the addition reaction was extremely rapid and 4 was obtained.



References and Notes.

1. (a) Ionin, B.I.; Petrov, A.A. J. Gen. Chem. USSR 1965, 35, 1910. (b) Our yields of 2 are considerably higher than those reported by Ionin and Petrov, see: Kyba, E.P.; Rines, S.P.; Owens, P.W.; Chou, S.-S.P. Tetrahedron Lett. 1981, 22, 1875.
2. Although this communication deals only with species derived from 1, other, more highly functionalized phosphorylalkynes behave analogously.
3. All spectroscopic and analytical properties were in accord with the assigned structures.
4. In the case of the methylamine adduct 3 ( $R^1 = H$ ,  $R = Me$ ), for example, the  $^1H$  NMR spectrum featured a doublet of quartets at  $\delta 7.1$  ( $NH$ ,  $^3J_{P-H} = 33$  Hz,  $^3J_{H-H} = 5$  Hz) and a doublet at 3.0 ppm ( $N-CH_3$ ,  $^3J_{H-H} = 5$  Hz); the former disappears and the latter collapses to a singlet upon addition of  $D_2O$  (the exchange is slow, ca. 15h).
5. Whitesell, J.K.; Whitesell, M.A. Synthesis 1983, 0000.
6. Compounds prepared, yields ( $^{31}P$  NMR in  $CDCl_3$ ): 7b ( $R^2 = Me$ ), 94%, two isomers, 4:1 ratio [major isomer, +26.7(d,  $J = 19$  Hz), +1.7(d,  $J = 19$  Hz); minor isomer, +25.0(d,  $J = 13$  Hz), +5.3 ppm(d,  $J = 13$  Hz)]; 7b ( $R^2 = allyl$ ), 90%; two isomers, 3:2 ratio [minor isomer, +27.7(d,  $J = 7$  Hz), +0.1 ppm(d,  $J = 7$  Hz), major isomer, +25.2(d,  $J = 13$  Hz), +1.6 ppm(d,  $J = 13$  Hz)]; 7b ( $R^2 = benzyl$ ), 98%, two isomers, 2:3 ratio [major isomer, +27.3(d,  $J = 6$  Hz), -0.3 ppm(d,  $J = 6$  Hz); minor isomer, +25.0(d,  $J = 11$  Hz), +1.7 ppm(d,  $J = 11$  Hz)].
7. Compounds prepared, yields ( $^{31}P$  NMR in  $CDCl_3$ ): 9b ( $R^2 = i-Pr$ ), 68% (+7.58 ppm); 9b ( $R^2 = t-Bu$ ), 94% (+7.56 ppm); 9b ( $R^2 = Ph$ ), 81% (+7.18 ppm).
8. (a) Walker, B.J. In "Organophosphorus Reagents in Organic Reagents"; Cadogan, J.I.G., Ed.; Academic Press, New York, NY 1979; Chapter 3. (b) Nagata, W.; Hayase, Y. J. Chem. Soc., (C) 1969, 460.
9. (a) Berlin, K.D.; Clauch, R.T.; Gaudy, E.T.; J. Org. Chem. 1968, 33, 3090. (b) Berlin, K.D.; Roy, N.K.; Clauch, R.T.; Bude, D. J. Am. Chem. Soc. 1968, 90, 4495. (c) Kowalik, J.; Kupczyk-Subotkowska, L.; Mastalerz, P. Synthetis 1981, 57. (d) Roberts,

E.; Simonsen, D.G.; Horiguchi, M.; Kittredge, J.S. Science **1968**, 159, 886. (e) Quinn, L.D. Ibid. **1964**, 144, 1134.

Table. Yields and  $^{31}\text{P}$  NMR Data<sup>a</sup> for the Production of Compounds 3 and 4

|                                     | 3a    | 3b       | 3c                    | 4d <sup>b</sup>   | 3e                              | 4f <sup>c</sup> |
|-------------------------------------|-------|----------|-----------------------|-------------------|---------------------------------|-----------------|
| R, R <sup>1</sup>                   | H, Me | H, c-Hex | H, CH <sub>2</sub> Ph | R=NH <sub>2</sub> | (CH <sub>2</sub> ) <sub>4</sub> | R=OH            |
| Yield                               | 93%   | 93%      | 100                   | 100               | 100                             | 85              |
| $^{31}\text{P}$ NMR: $\delta$ (ppm) | +16.7 | +17.14   | +16.4                 | +17.3             | +16.7                           | +15.35          |
| $J_{\text{AB}}$ (ppm)               | 2.7   | 2.7      | 2.6                   | 0.50              | 2.5                             | 0.28            |
| $\nu_{\text{AB}}$ (ppm)             | 11.7  | 11.3     | 11.3                  | 12.7              | 8.6                             | 13.4            |

<sup>a</sup> Spectra determined on ca. 0.1M solutions in CDCl<sub>3</sub> at 32.2MHz and ambient temperature.

<sup>b</sup> Only the hydrazone tautomer observed.

<sup>c</sup> Only the oxime tautomer observed.

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